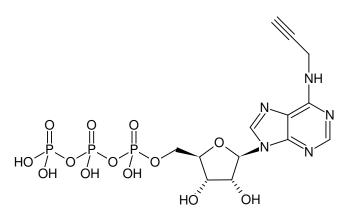




N⁶-Propargyl-ATP (N⁶pATP)

N⁶-Propargyl-adenosine-5'-triphosphate, Sodium salt

| Cat. No. | Amount |
|--------------|--------|
| CLK-NU-001-1 | 1 mg |
| CLK-NU-001-5 | 5 mg |



Structural formula of N⁶-Propargyl-ATP (N⁶pATP)

For general laboratory use.

Shipping: shipped on gel packs

Storage Conditions: store at -20 °C

Short term exposure (up to 1 week cumulative) to ambient temperature possible.

Shelf Life: 12 months after date of delivery

Molecular Formula: C₁₃H₁₈N₅O₁₃P₃ (free acid)

Molecular Weight: 545.23 g/mol (free acid)

Exact Mass: 545.01 g/mol (free acid)

Purity: ≥ 95 % (HPLC)

Form: solid

Color: white to off-white

Solubility: 10 mM Tris-HCl pH 7.5

Spectroscopic Properties: λ_{max} 262 nm, ϵ 18.0 L mmol⁻¹ cm⁻¹ (Tris-HCl pH 7.5)

Applications:

in vitro AMPylation of proteins^[1,2]

in vitro polyadenylation of RNA^[3]

The resulting alkyne-functionalized protein^[1,2] or RNA^[3] can subsequently be processed via Cu(I)-catalyzed (azide-alkyne) click chemistry that offers the choice

- to introduce a Biotin group for subsequent purification tasks (via Azides of Biotin)
- to introduce fluorescent group for subsequent microscopic imaging (via Azides of fluorescent dyes)
- to crosslink the RNA to azide-functionalized biomolecules e.g.proteins

Presolski *et al.*^[4] and Hong *et al.*^[5] provide a general protocol for Cu(I)-catalyzed click chemistry reactions that may be used as a starting point for the set up and optimization of individual assays.

Agonistic ligand, mainly for nucleoside receptor A1

Nucleoside-triphosphates can be converted by different membranebound phosphatases into nucleosides acting as nucleoside receptor ligands. In some cases nucleoside phosphates act also directly on nucleoside receptors.

Please note: This compound contains a phosphoramide linkage which is hydrolyzed at pH <7.0.

For preparation of a 10 mM solution use 100 mM buffer (for example: bicarbonate, borate, phosphate and Tris) to prevent degradation at acidic pH.

Related Products:

Copper (II)-Sulphate (CuSO₄), #CLK-MI004 Tris(3-hydroxypropyltriazolylmethyl)amine (THPTA), #CLK-1010 Sodium Ascorbate (Na-Ascorbate), #CLK-MI005

Selected References:

[1] Grammel *et al.* (2011) A Chemical Reporter for Protein AMPylation. *J. Am. Chem. Soc.* **133**:17103.

[2] Broncel *et al.* (2012) A New Chemical Handle for Protein AMPylation at the Host-Pathogen Interface. *ChemBioChem* **13**:183.

[3] Grammel *et al.* (2012) Chemical Reporter for Monitoring RNA Synthesis and Poly (A) Tail Dynamics. *ChemBioChem* **13**:1112.

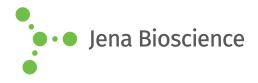
[4] Presolski et al. (2011) Copper-Catalyzed Azide-Alkyne Click Chemistry for Bioconjugation. *Current Protocols in Chemical Biology* **3**:153.

[5] Hong *et al.* (2011) Analysis and Optimization of Copper-Catalyzed Azide-Alkyne Cycloaddition for Bioconjugation. *Angew. Chem. Int. Ed.* **48**:9879.

Sirci *et al.* (2012) Ligand-, structure- and pharmacophore-based molecular fingerprints: a case study on adenosine A1, A2A, A2B, and A3 receptor









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Joshi et al. (2005) Purine derivatives as ligands for A3 adenosine receptors. *Current Topics in Medicinal Chemistry* **5**:1275.

Volpini *et al.* (2002) N6-Alkyl-2-alkynyl derivatives of adenosine as potent and selective agonists of the human adenosine A3 receptor and starting point for searching A2B ligands. *J. Med. Chem.* **45 (15)**:3271.

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